

Bis[2,4-bis(2-pyridyl- $\kappa$ N)-6-(4-pyridyl)-1,3,5-triazine- $\kappa$ N<sup>3</sup>]ruthenium(II) bis(hexafluorophosphate)Richard M. Hartshorn,\*  
Ramin Zibaseresht and  
Ward T. RobinsonDepartment of Chemistry, University of  
Canterbury, Private Bag 4800, Christchurch,  
New ZealandCorrespondence e-mail:  
richard.hartshorn@canterbury.ac.nz

## Key indicators

Single-crystal X-ray study  
 $T = 85$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.008$  Å  
 $R$  factor = 0.058  
 $wR$  factor = 0.137  
Data-to-parameter ratio = 13.5For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

The coordination of Ru in the title compound,  $[\text{Ru}(\text{C}_{18}\text{H}_{12}\text{N}_6)_2](\text{PF}_6)_2$ , is octahedral. The coordination sphere is formed by six N atoms from the tridentate sites of two ligands that are meridionally oriented around the  $\text{Ru}^{\text{II}}$  ion. The locations of the donor atoms deviate substantially from an ideal octahedron. The Ru–N distances are in the range 1.983 (4)–2.111 (4) Å. The Ru atom occupies a special position with twofold rotational symmetry. One of the P atoms occupies a special position with twofold rotational symmetry and the other P atom is located on a special position at a centre of symmetry.

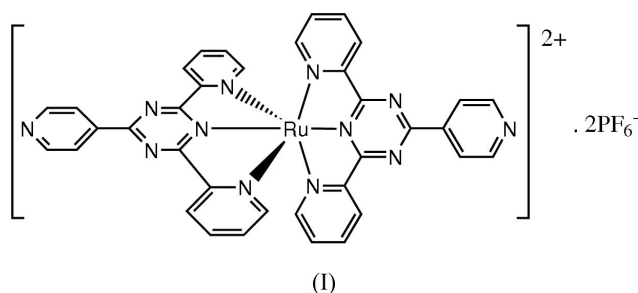
Received 15 April 2005

Accepted 20 April 2005

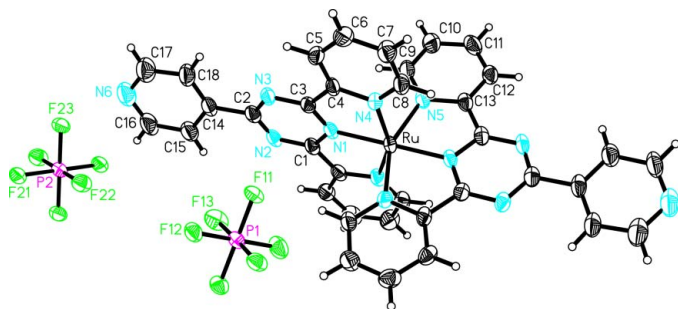
Online 27 April 2005

## Comment

Our aim involves the synthesis and characterization of bridging ligands containing two non-equivalent binding sites. Our interests are particularly focused on preparation of such ligands capable of binding an  $\text{Ru}^{\text{II}}$  ion at one site and a  $\text{Co}^{\text{III}}$  ion at the other site. As part of our study, the ligand 2,4-bis(2-pyridyl)-6-(4-pyridyl)-1,3,5-triazine was synthesized using the method of Polson *et al.* (2002). This terpyridyl-like ligand, which contains a remote fourth N-donor atom, was a candidate for our study. The title ruthenium complex, (I), in which two ligands are coordinated to an  $\text{Ru}^{\text{II}}$  ion through their tridentate sites was prepared according to the reported methods (Polson *et al.*, 2002, 2004).



Complex (I) (Fig. 1) crystallizes in the orthorhombic space group  $Pcca$  with four well separated molecules in the unit cell. The asymmetric unit comprises one half-cation and two half-anions. The Ru atom occupies a special position with twofold rotational symmetry. One of the P atoms occupies a special position with twofold rotational symmetry and the other P atom is located on a special position at a centre of symmetry. The ruthenium ion is six-coordinate, but the geometry is significantly distorted from that of an ideal octahedron as a result of the constraints enforced by the ligand structure. This is commonly observed for terpyridyl-like ligands (Thummel &


**Figure 1**

A perspective view of the Ru<sup>II</sup> complex (I), showing the atom-labelling scheme with 50% probability displacement ellipsoids. Unlabelled atoms in the cation and PF<sub>6</sub><sup>-</sup> anions P<sub>1</sub> and P<sub>2</sub> are related by the symmetry codes  $(\frac{1}{2} - x, -y, z)$ ,  $(\frac{1}{2} - x, 1 - y, z)$  and  $(1 - x, 2 - y, -z)$ , respectively. H atoms are drawn as small spheres of arbitrary radii.

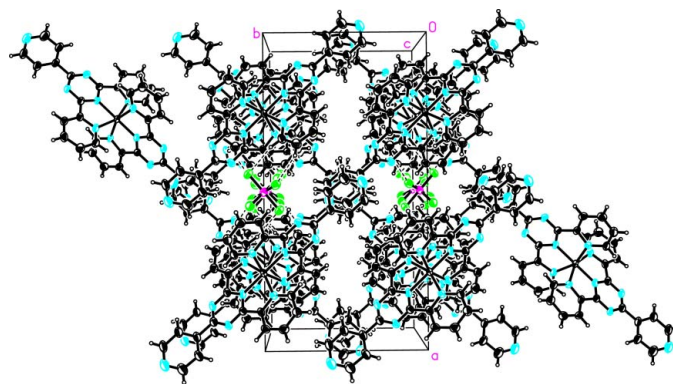
Jahng, 1986; Constable *et al.*, 1990, 1992, 2001; Cathey *et al.*, 1990; Bushell *et al.*, 1998; Craig *et al.*, 1998; Sasaki *et al.*, 1998; Cardenas *et al.*, 1999; Chamchoumis & Potvin, 1999; Fallahpour *et al.*, 1999; Lashgari *et al.*, 1999; Pyo *et al.*, 1999; Ziegler *et al.*, 1999; Alcock *et al.*, 2000; Encinas *et al.*, 2002; Fang *et al.*, 2002; Laine *et al.*, 2002). Each of the tridentate ligands occupies three meridionally located coordination sites. The Ru–N bond lengths show the expected pattern for complexes related to [Ru(terpy)<sub>2</sub>](PF<sub>6</sub>)<sub>2</sub> (Pyo *et al.*, 1999; Lashgari *et al.*, 1999), with short metal–ligand bonds to the central ring [1.983 (4) Å] and longer bonds to the side pyridine rings [mean 2.108 (4) Å]. The non-coordinated pyridine rings are not coplanar with the central triazine ring and form interannular angles of 16.8 (3)°. Similar observations have been made for the Ni<sup>II</sup> and Ru<sup>II</sup> complexes based on the 4'-phenyl-substituted terpyridine system (Constable *et al.*, 1990; Chamchoumis *et al.*, 1999). As shown in Fig. 2, there are  $\pi$ – $\pi$  stacking interactions (face–face) between the terminal pyridyl rings of adjacent complexes [centroid–centroid distance = 3.72 (8) Å]. The complex cations are arranged like the steps of a spiral staircase along the *b* axis of the unit cell, with the stack of pyridyl rings forming the pole of the staircase. There appears to be no significant hydrogen bonding in the crystal structure.

## Experimental

The 2,4-bis(2-pyridyl)-6-(4-pyridyl)-1,3,5-triazine ligand and its ruthenium complex, (I), were synthesized according to the methods of Polson *et al.* (2002, 2004). Red block-shaped crystals were obtained by slow evaporation of a CH<sub>3</sub>CN–H<sub>2</sub>O solution of the complex (yield 16%).

### Crystal data

|  |                                       |
|--|---------------------------------------|
| [Ru(C <sub>18</sub> H <sub>12</sub> N <sub>6</sub> ) <sub>2</sub> ](PF <sub>6</sub> ) <sub>2</sub> | Mo K $\alpha$ radiation               |
| <i>M<sub>r</sub></i> = 1015.68   | Cell parameters from 8243 reflections |
| Orthorhombic, <i>Pcca</i>  | $\theta$ = 4.9–52.5°                  |
| <i>a</i> = 21.474 (11) Å   | $\mu$ = 0.61 mm <sup>-1</sup>         |
| <i>b</i> = 11.008 (6) Å  | <i>T</i> = 85 (2) K                   |
| <i>c</i> = 16.044 (8) Å  | Block, red                            |
| <i>V</i> = 3793 (3) Å <sup>3</sup>   | 0.61 × 0.29 × 0.08 mm                 |
| <i>Z</i> = 4   |                                       |
| <i>D<sub>x</sub></i> = 1.779 Mg m <sup>-3</sup>  |                                       |


**Figure 2**

A packing diagram, showing the  $\pi$ – $\pi$  stacking interactions between the complexes parallel to the *ac* plane.

### Data collection

|  |  |
|--|--|
| Bruker SMART CCD diffractometer                          | 3864 independent reflections           |
| $\varphi$ and $\omega$ scans                             | 2935 reflections with $I > 2\sigma(I)$ |
| Absorption correction: multi-scan (SADABS; Bruker, 1999) | $R_{\text{int}} = 0.044$               |
| $T_{\text{min}} = 0.709$ , $T_{\text{max}} = 0.953$      | $\theta_{\text{max}} = 26.4^\circ$     |
| 21 050 measured reflections                              | $h = -26 \rightarrow 26$               |
|  | $k = -13 \rightarrow 11$               |
|  | $l = -18 \rightarrow 19$               |

### Refinement

|                                 |  |
|---------------------------------|--|
| Refinement on $F^2$             | $w = 1/[\sigma^2(F_o^2) + (0.0303P)^2 + 19.9599P]$   |
| $R[F^2 > 2\sigma(F^2)] = 0.058$ | where $P = (F_o^2 + 2F_c^2)/3$                       |
| $wR(F^2) = 0.137$               | $(\Delta/\sigma)_{\text{max}} < 0.001$               |
| $S = 1.16$                      | $\Delta\rho_{\text{max}} = 1.54 \text{ e \AA}^{-3}$  |
| 3864 reflections                | $\Delta\rho_{\text{min}} = -1.09 \text{ e \AA}^{-3}$ |
| 287 parameters                  |  |
| H-atom parameters constrained   |  |

**Table 1**

Selected geometric parameters (Å, °).

|                                     |             |                                     |             |
|-------------------------------------|-------------|-------------------------------------|-------------|
| Ru–N1                               | 1.983 (4)   | Ru–N5                               | 2.111 (4)   |
| Ru–N4                               | 2.104 (4)   |                                     |             |
| N1 <sup>i</sup> –Ru–N1              | 178.9 (3)   | N1–Ru–N5 <sup>i</sup>               | 77.41 (18)  |
| N1 <sup>i</sup> –Ru–N4 <sup>i</sup> | 77.00 (17)  | N4 <sup>i</sup> –Ru–N5 <sup>i</sup> | 92.97 (17)  |
| N1–Ru–N4 <sup>i</sup>               | 102.22 (16) | N4–Ru–N5 <sup>i</sup>               | 154.38 (16) |
| N1–Ru–N4                            | 77.00 (17)  | N1 <sup>i</sup> –Ru–N5              | 77.41 (18)  |
| N4 <sup>i</sup> –Ru–N4              | 93.6 (2)    | N5 <sup>i</sup> –Ru–N5              | 91.7 (2)    |
| N1 <sup>i</sup> –Ru–N5 <sup>i</sup> | 103.38 (17) |                                     |             |
| N3–C2–C14–C18                       | 16.9 (8)    | N2–C2–C14–C15                       | 16.2 (8)    |

Symmetry code: (i)  $\frac{1}{2} - x, -y, z$ .

All H atoms were placed in calculated positions [C–H = 0.93 Å and  $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ ] and were included in the refinement in the riding-model approximation. The highest peak is very close to Ru (0.89 Å) and the deepest hole is near N4 (1.62 Å), neither in chemically sensible locations for atoms.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT-Plus (Bruker, 1999); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXTL (Sheldrick, 2001); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

RZ would like to acknowledge receipt of a University of Canterbury doctoral scholarship.

## References

- Alcock, N. W., Barker, P. R., Haider, J. M., Hannon, M. J., Painting, C. L., Pikramenou, Z., Plummer, E. A., Rissanen, K. & Saarenketo, P. (2000). *J. Chem. Soc. Dalton Trans.* pp. 1447–1462.
- Bruker (1999). *SADABS, SAINT-Plus* (Version 6.22) and *SMART* (Version 5.045). Bruker AXS Inc., Madison, Wisconsin, USA.
- Bushell, K. L., Couchman, S. M., Jeffery, J. C., Rees, L. H. & Ward, M. D. (1998). *J. Chem. Soc. Dalton Trans.* pp. 3397–3404.
- Cardenas, D. J., Collin, J.-P., Gavina, P., Sauvage, J.-P., De Cian, A., Fischer, J., Armaroli, N., Flamigni, L., Vicinelli, V. & Balzani, V. (1999). *J. Am. Chem. Soc.* **121**, 5481–5488.
- Cathey, C. J., Constable, E. C., Hannon, M. J., Tocher, D. A. & Ward, M. D. (1990). *J. Chem. Soc. Chem. Commun.* pp. 621–622.
- Chamchoumis, C. M. & Potvin, P. G. (1999). *J. Chem. Soc. Dalton Trans.* pp. 1373–1374.
- Constable, E. C., Housecroft, C. E., Johnston, L. A., Armspach, D., Neuburger, M. & Zehnder, M. (2001). *Polyhedron*, **20**, 483–492.
- Constable, E. C., Lewis, J., Liptrot, M. C. & Raithby, P. R. (1990). *Inorg. Chim. Acta*, **178**, 47–54.
- Constable, E. C., Thompson, A. M. W. C., Tocher, D. A. & Daniels, M. A. M. (1992). *New J. Chem.* **16**, 855–67.
- Craig, D. C., Scudder, M. L., McHale, W.-A. & Goodwin, H. A. (1998). *Aust. J. Chem.* **51**, 1131–1139.
- Encinas, S., Flamigni, L., Barigelletti, F., Constable, E. C., Housecroft, C. E., Schofield, E. R., Figgemeier, E., Fenske, D., Neuburger, M., Vos, J. G. & Zehnder, M. (2002). *Chem. Eur. J.* **8**, 137–150.
- Fallahpour, R.-A., Neuburger, M. & Zehnder, M. (1999). *Polyhedron*, **18**, 2445–2454.
- Fang, Y.-Q., Taylor, N. J., Hanan, G. S., Loiseau, F., Passalacqua, R., Campagna, S., Nierengarten, H. & Van Dorsselaer, A. (2002). *J. Am. Chem. Soc.* **124**, 7912–7913.
- Laine, P., Bedioui, F., Ochsenbein, P., Marvaud, V., Bonin, M. & Amouyal, E. A. (2002). *J. Am. Chem. Soc.* **124**, 1364–1377.
- Lashgari, K., Norrestam, R. & Akermark, B. (1999). *Acta Cryst.* **C55**, IUC9900108.
- Polson, M. I. J., Medlycott, E. A., Hanan, G. S., Mikelsons, L., Taylor, N. J., Watanabe, M., Tanaka, Y., Loiseau, F., Passalacqua, R. & Campagna, S. (2004). *Chem. Eur. J.* **10**, 3640–3648.
- Polson, M. I. J., Taylor, N. J. & Hanan, G. S. (2002). *Chem. Commun.* pp. 1356–1357.
- Pyo, S., Perez-Cordero, E., Bott, S. G. & Echegoyen, L. (1999). *Inorg. Chem.* **38**, 3337–3343.
- Sasaki, I., Daran, J. C., Ait-Haddou, H. & Balavoine, G. G. A. (1998). *Inorg. Chem. Commun.* **1**, 354–357.
- Sheldrick, G. M. (2001). *SHELXTL*. Version 6.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Thummel, R. P. & Jahng, Y. (1986). *Inorg. Chem.* **25**, 2527–2534.
- Ziegler, M., Monney, V., Stoeckli-Evans, H., Von Zelewsky, A., Sasaki, I., Dupic, G., Daran, J.-C. & Balavoine, G. G. A. (1999). *J. Chem. Soc. Dalton Trans.* pp. 667–676.